



University of Natural Resources and
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Department of
Water, Atmosphere, and Environment

Modelling in Sanitary Engineering

811.360 VU 3.0

Wastewater treatment plant – Part 1

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9 November 2011



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Modelling of WWTPs

Content

- Theory:
 - **Components of a WWTP model (Flow and transport models, Biokinetic models, Activated Sludge Models, Settler models)**
 - Control of AS plants (incl. sensor models)
 - Simulation studies (Unified Protocol data quality, calibration)
 - Simulation of constructed wetlands for wastewater treatment
- Modelling exercises:
 - **Development of a biokinetic model (nitrification)**
 - **Dynamic simulation of a AS plant**
 - Control of a AS plant
 - Calibration of a WWTP model

Components of a WWTP model

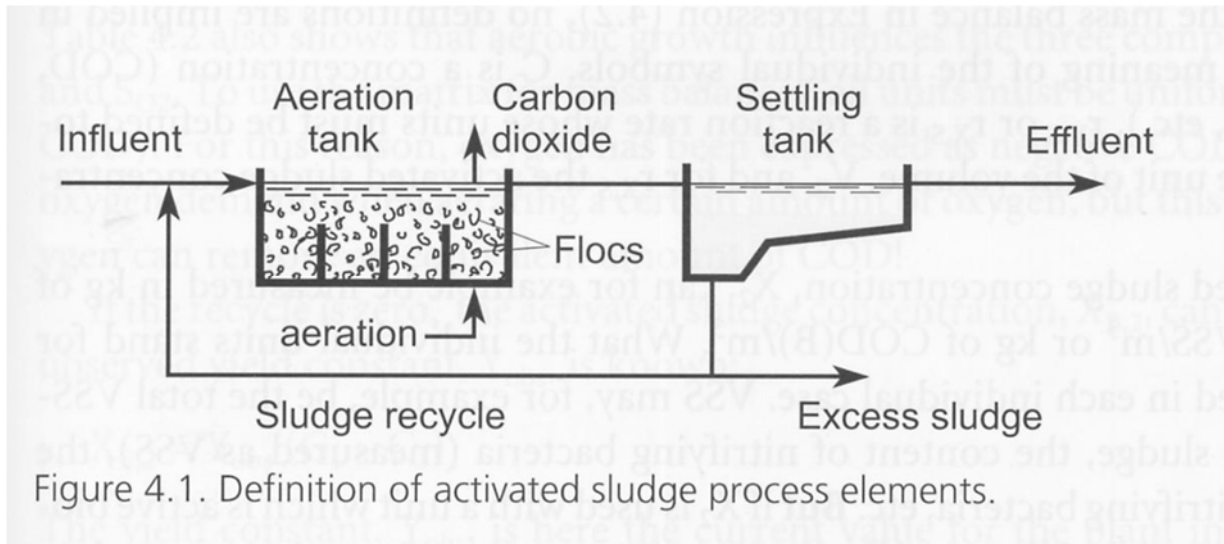
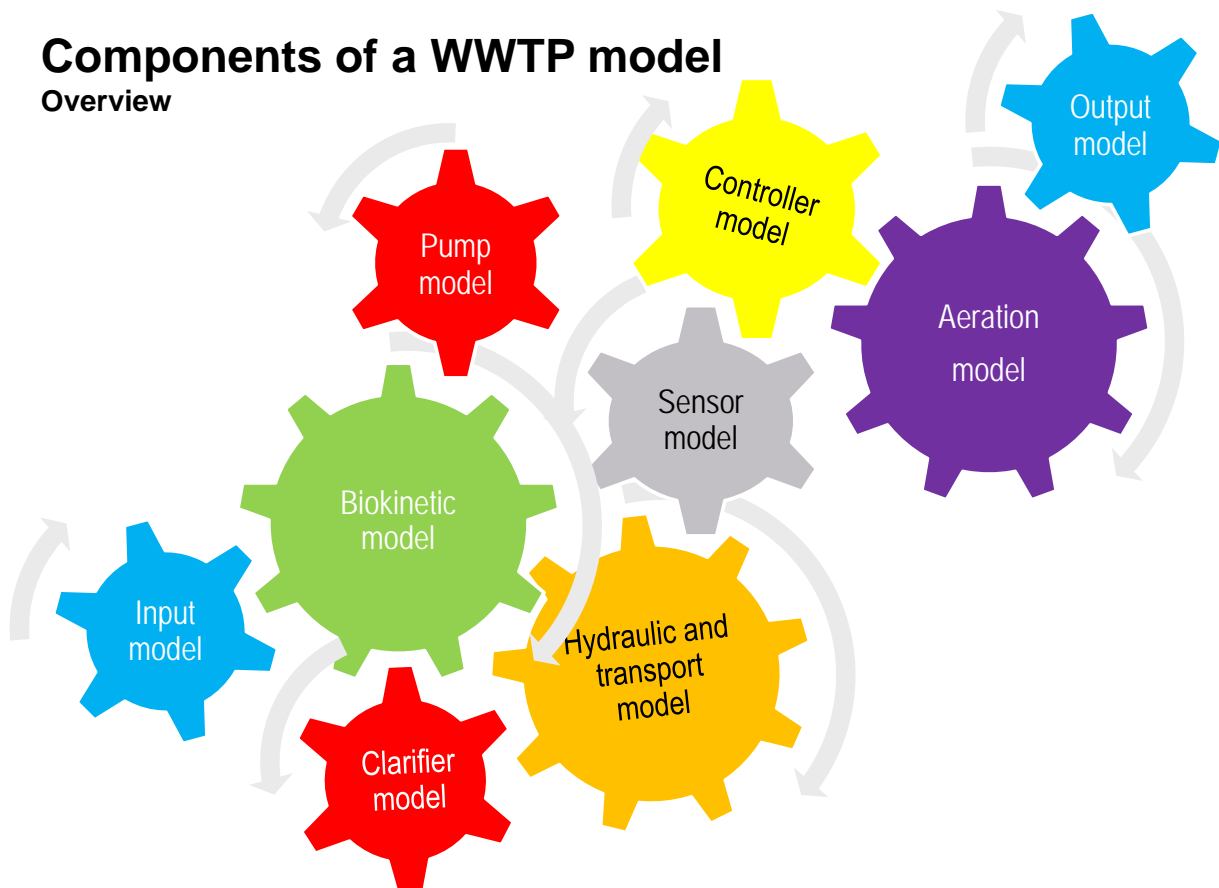


Figure 4.1. Definition of activated sludge process elements.

Components of a WWTP model

Overview



Components of a WWTP model

Overview

- Flow model (hydraulics)
- Model of transport processes
 - Settling: Primary and secondary clarifier
 - Activated sludge tank
 - Distribution and mixing devices
- Biokinetic model (model of the reactions)
- Operation parameters (control, sensors/actuators such as pumps and aeration)
- Input and output models

Components of a WWTP model

Flow and transport models

- Ideal reactors
- Transport processes
- Tracer experiments

An accurate description of the flow and transport processes is at least as important as the biokinetic model

Components of a WWTP model

Flow and transport models

Ideal reactors

- Completely stirred tank reactor – CSTR
- Cascade of CSTRs
- CSTR with variable volume
- Plug-flow reactor

Components of a WWTP model

Flow and transport models - Ideal reactors

Completely stirred tank reactor – CSTR

Properties of a CSTR:

- Inflow = outflow
- Volume = constant
- Mixing homogeneous
- No concentration gradients
- Effluent concentration = concentration in the reactor
- Usually in steady-state

$$\text{Mass balance: } V \cdot \frac{dC_A}{dt} = Q \cdot C_{A,in} - Q \cdot C_A + r_A \cdot V$$

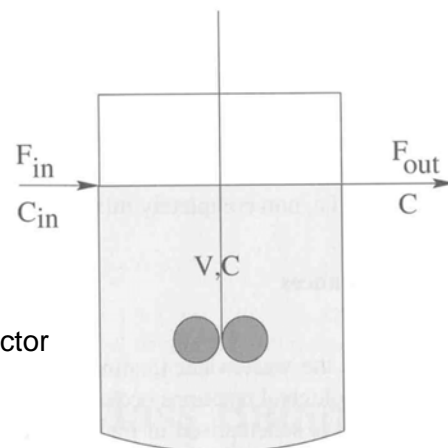
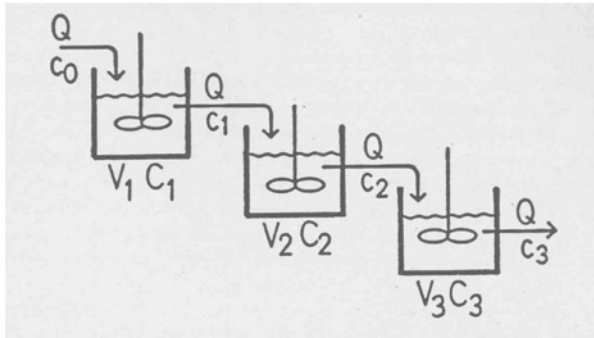


FIG. 2.1. Schematic view of a stirred tank reactor.

Components of a WWTP model

Flow and transport models - Ideal reactors

Cascade of CSTRs



$$\frac{C_k}{C_{k-1}} = \frac{1}{1 + K \cdot V_k / Q} \quad \text{with } r = -K \cdot C_k$$

$$\frac{C_k}{C_0} = \frac{1}{\left(1 + \frac{K \cdot V_{ges}}{k \cdot Q}\right)^k} \quad \text{with } V_{ges} = \sum_k V_k \quad \text{and } V_k = \frac{V_{ges}}{k}$$

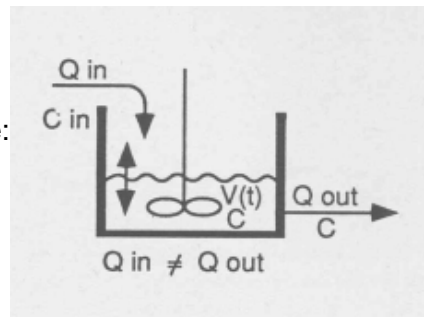
Components of a WWTP model

Flow and transport models - Ideal reactors

CSTR with variable volume

Properties of CSTRs with variable volume:

- Inflow not equal outflow
- Volume variable
- Mixing homogeneous
- No concentration gradients
- Effluent concentration = concentration in the reactor
- Steady-state corresponds to CSTR
- Open system



$$\frac{dV}{dt} = Q_{in} - Q_{out}$$

$$\frac{d(V \cdot C_A)}{dt} = C_A \cdot \frac{dV}{dt} + V(t) \cdot \frac{dC_A}{dt} = Q_{in} \cdot C_{A,in} - Q_{out} \cdot C_A + r_A \cdot V(t)$$

$$V(t) \cdot \frac{dC_A}{dt} = Q_{in} \cdot (C_{A,in} - C_A) + r_A \cdot V(t)$$

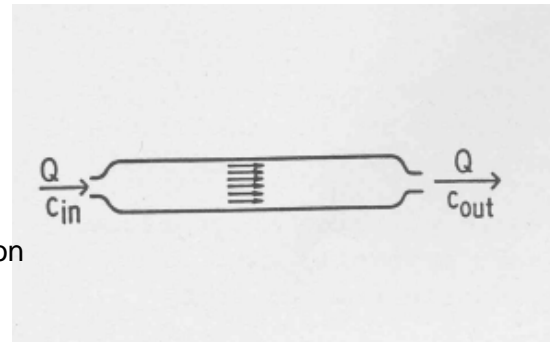
Components of a WWTP model

Flow and transport models - Ideal reactors

Plug-flow reactor

Properties of plug-flow reactors:

- Inflow = outflow
- Volume constant
- Homogeneous Mixing only in dx
- Concentration gradients along flow direction
- Steady-state is interesting



$$\frac{\partial C_A}{\partial t} = -\frac{Q}{F} \cdot \frac{\partial C_A}{\partial x} + r_A$$

Components of a WWTP model

Flow and transport models

Transport processes

- Convection (advection or turbulence)
- Molecular diffusion
- Sedimentation / Floatation
- Dispersion

Components of a WWTP model

Flow and transport models - Transport processes

Convection

- used synonymously with advection
- transport of migrants due to bulk flow of a fluid mixed phase
- $j_A = v \cdot C_A$ (mass flux density)
- v ... flow velocity (vector)

- in turbulent water no analytical description is possible

Components of a WWTP model

Flow and transport models - Transport processes

Molecular diffusion

- Primarily due to Brownian motion in solids, liquids, and gases
- Transport von molecules driven by concentration gradients (in the direction from higher concentrations to lower concentrations)
- Fick's 1st law
- $j_{A,x} = -D_A \cdot dC_A/dx$ (1D mass flux density)
- D_A ... diffusions coefficient of migrant A in liquid phase

Components of a WWTP model

Flow and transport models - Transport processes

Sedimentation / Flotation

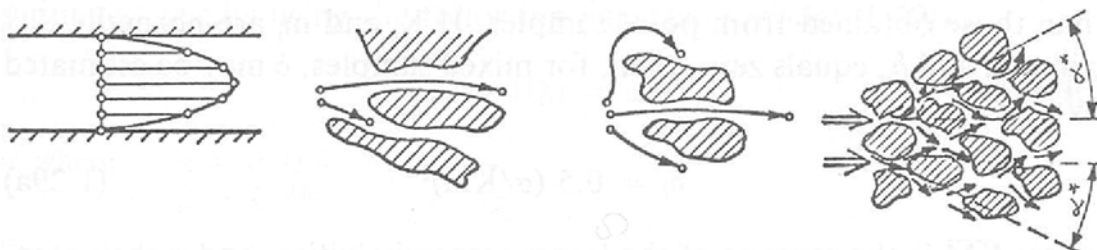
- Transport due to gravity
- $j_A = v_S \cdot C_A$ (mass flux density)
- v_S ... sedimentation velocity
- Flotation: opposite direction

Components of a WWTP model

Flow and transport models - Transport processes

Hydrodynamic dispersion

- Motion of migrants relative to the convective motion (bulk flow)
- Mathematically described as diffusion



Components of a WWTP model

Flow and transport models - Tracer experiments

Tracer experiments

- To measure the HRT of systems

Properties of tracer substances

- the same transport properties as water (similar molecular diffusion, no sedimentation)
- no influence on mixing and flow (temperature, density)
- no reaction and no adsorption
- cheap, high solubility in water, and easy to analyse
- background concentration low and/or constant

Components of a WWTP model

Flow and transport models - Tracer experiments

Possible tracers

- Salts
 - NaCl und KCl can be measured with the electrical conductivity
 - Li⁺ can be measured with atomic adsorption
 - Br⁻ low background concentration
- fluorescent tracers
- radioactive tracers
 - very accurate, but radioactive (usually special permission needed)

Components of a WWTP model

Biokinetic models

describe

- **Degradation of organic matter**
 - Oxygen consumption and sludge production
- **Nutrient elimination**
 - Nitrogen (Nitrification/Denitrification)
 - Phosphorus (Bio-P, Precipitation)

Components of a WWTP model

Biokinetic models

Development of a biokinetic model for nitrification

Nitrification



14 g NH₄-N require 64 g O₂

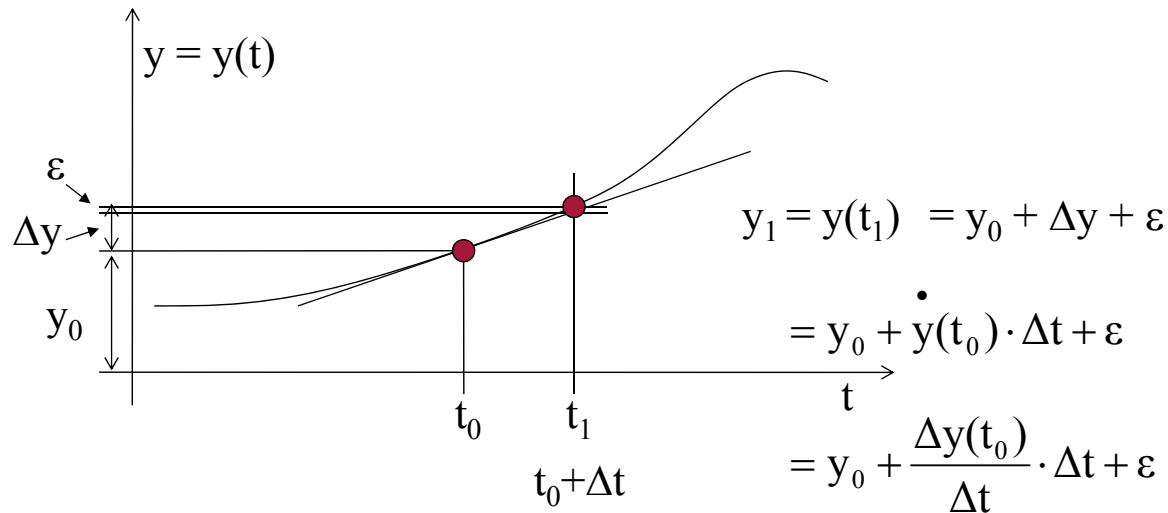
- 4,57 g O₂/g NH₄-N and +1 g NO₃-N/g NH₄-N

2-stage process:



In reality much more complicated.

Numerical integration



Components of a WWTP model

Development of a biokinetic model for nitrification

Model 1 - Reaction with zero-order kinetics

Reaction rate: $r = k_0$

1 g $\text{NH}_4\text{-N}$ result in 1 g $\text{NO}_3\text{-N}$

100 mg XAUT/l

$k_0 = 11 \text{ g}_\text{N} / \text{h}$

Excel-File: Nitrification.xls[Model 1]

Components of a WWTP model

Development of a biokinetic model for nitrification

Model 2 - Reaction with first-order kinetics

Reaction rate: $r = k_1 \cdot C_{\text{SNH}_4}$
 $k_1 = 0.15 \text{ g}_N / (\text{g}_N \cdot \text{h})$

Components of a WWTP model

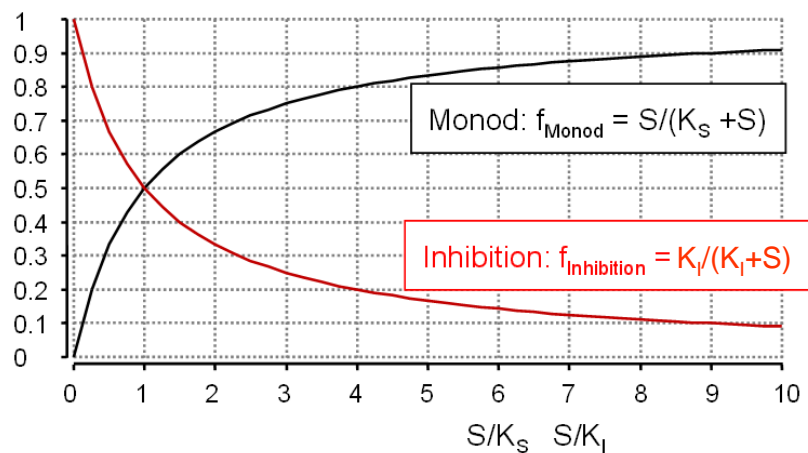
Development of a biokinetic model for nitrification

Model 3 - Reaction with Monod kinetics

Growth rate $\mu = \mu_{\text{max}} \cdot f_{\text{Monod}}$
 Reaction rate $r = \mu \cdot X$

Stoichiometric matrix related
to biomass

$\mu_m = 0.15 \text{ g}_{\text{NO}_3\text{-N}} / (\text{g}_{\text{COD, BM}} \cdot \text{h})$
 $k_S = 0.20 \text{ mg}_{\text{NH}_4\text{-N}} / \text{l}$



Components of a WWTP model

Development of a biokinetic model for nitrification

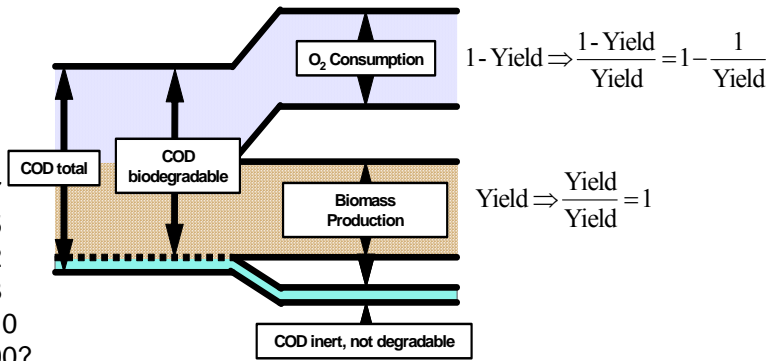
Model 4 - including bacterial growth

Autotrophic yield

- $Y_{AUT} = 0.24 \text{ g}_{COD, BM} / \text{g}_N$

Yield for different biological systems

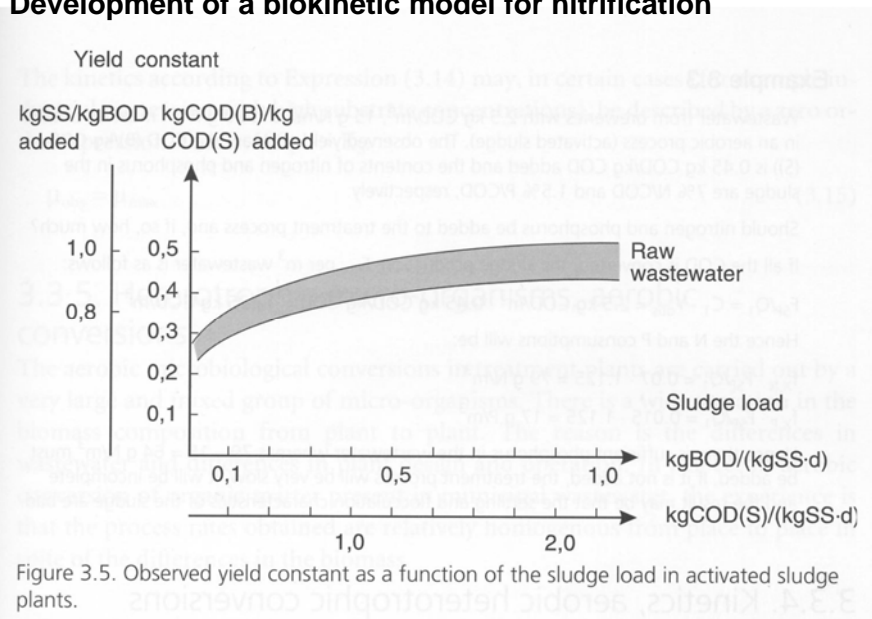
- Heterotrophic bacteria 0.67
- Fish (1 yr, 0.5 kg) 0.45
- Hen 0.32
- Pig (to 65 kg) 0.23
- Human (0-16 yrs) 0.010
- Human (from 16 yrs) 0.000?



(in kg body weight produced per kg food consumed, Henze et al., 2002)

Components of a WWTP model

Development of a biokinetic model for nitrification



Components of a WWTP model

Development of a biokinetic model for nitrification

Model 4a - including N incorporation (organic N of biomass)

$$i_N = 0.08 \text{ g}_N / \text{g}_{\text{COD, BM}}$$

Components of a WWTP model

Development of a biokinetic model for nitrification

Model 5 - including microbial decay (lysis)

Concepts to describe the energy balance of micro-organisms

- Endogenous respiration
 - direct respiration (oxygen demand) of micro-organisms (O_2 or NO_3^- required)
- Death-regeneration (lysis)
 - Death \rightarrow degradation \rightarrow new substrate is generated that is available for other micro-organisms [biomass \rightarrow substrate (COD) + nutrients (N + P)]
 - Results in indirect oxygen demand

$$r = b \cdot X$$

$$b = 0.025 \text{ g}_{\text{COD}} / (\text{g}_{\text{COD, BM}} \cdot \text{h})$$

Components of a WWTP model

Development of a biokinetic model for nitrification

Model 6 - including oxygen consumption

$$+ 1 \text{ g}_{\text{O}_2} / \text{g}_{\text{COD, BM}} \text{ and } - 4.57 \text{ g}_{\text{O}_2} / \text{g}_{\text{NH}_4\text{-N}}$$

$$\frac{4.57 / 0.24}{(\text{g}_{\text{O}_2} / \text{g}_{\text{NH}_4\text{-N}}) / (\text{g}_{\text{COD, BM}} / \text{g}_{\text{NH}_4\text{-N}})}$$

Stoichiometric factor $1 - \frac{4.57}{Y_{\text{AUT}}} = - \frac{4.57 - Y_{\text{AUT}}}{Y_{\text{AUT}}}$

Growth rate $\mu = \mu_{\text{max}} \cdot \frac{C_{\text{SNH}_4}}{K_{\text{S, NH}_4} + C_{\text{SNH}_4}} \cdot \frac{C_{\text{SO}_2}}{K_{\text{S, O}_2} + C_{\text{SO}_2}}$

$K_{\text{S, O}_2} = 1 \text{ mg}_{\text{O}_2} / \text{l}$

Components of a WWTP model

Development of a biokinetic model for nitrification

Model 7 - including aeration

$$\frac{dC_{\text{SO}_2}}{dt} = kLa \cdot (C_{\text{SO}_2, \text{sat}} - C_{\text{SO}_2})$$

$k_{\text{La}} = 0.1 \text{ h}^{-1}$

$C_{\text{SO}_2, \text{sat}} (20^\circ\text{C}) = 9.08 \text{ mg}_{\text{O}_2} / \text{l}$

Components of a WWTP model

Development of a biokinetic model for activated sludge plants

describing

- **Degradation of organic matter**
 - Oxygen consumption and sludge production
- **Nutrient elimination**
 - Nitrogen (Nitrification/Denitrification)
 - Phosphorus (Bio-P, Precipitation)

Components of a WWTP model

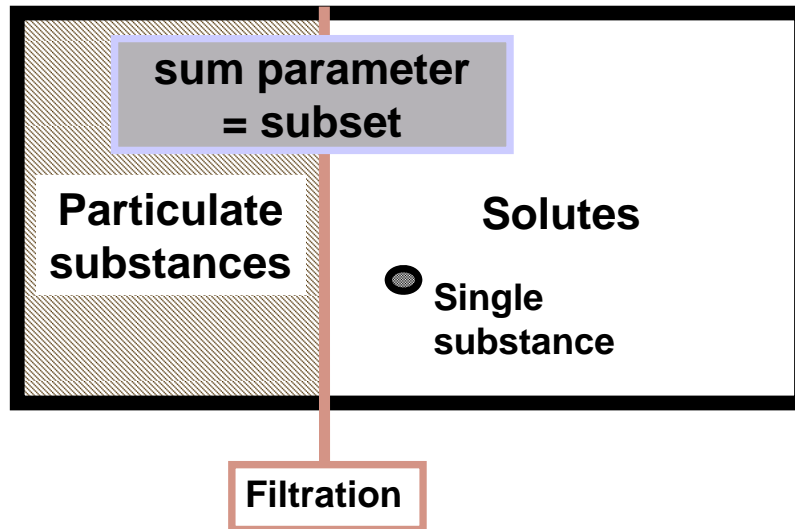
Development of a biokinetic model for AS plants

The concept of sum parameters

- Wastewater is a „raw product“ with unknown composition
- Numerous organic substances → not detectable as single substances
- Behaviour of sum parameters is described with models
- Behaviour of single substances (e.g. ammonia is a special case)

Components of a WWTP model

Development of a biokinetic model for AS plants

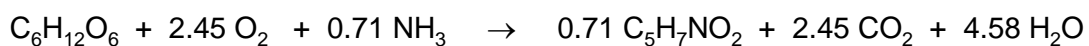


Components of a WWTP model

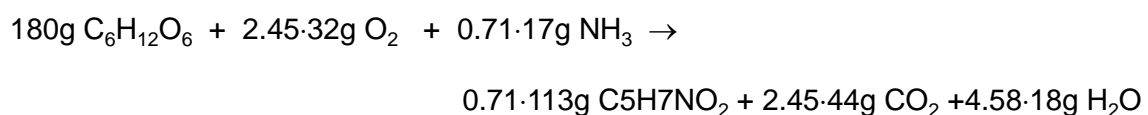
Development of a biokinetic model for AS plants

Example: Aerobic degradation of glucose (1)

in mol:



in mass units :



Components of a WWTP model

Development of a biokinetic model for AS plants

Example: Aerobic degradation of glucose (2)

Stoichiometric factors n based on biomass:

| | |
|--|---|
| reactants: | products: |
| - 2.24 g _{C₆H₁₂O₆} / g _{BM} | + 1 g _{C₅H₇NO₂} / g _{BM} |
| - 0.98 g _{O₂} / g _{BM} | + 1.34 g _{CO₂} / g _{BM} |
| - 0.15 g _{NH₃} / g _{BM} | + 1.03 g _{H₂O} / g _{BM} |

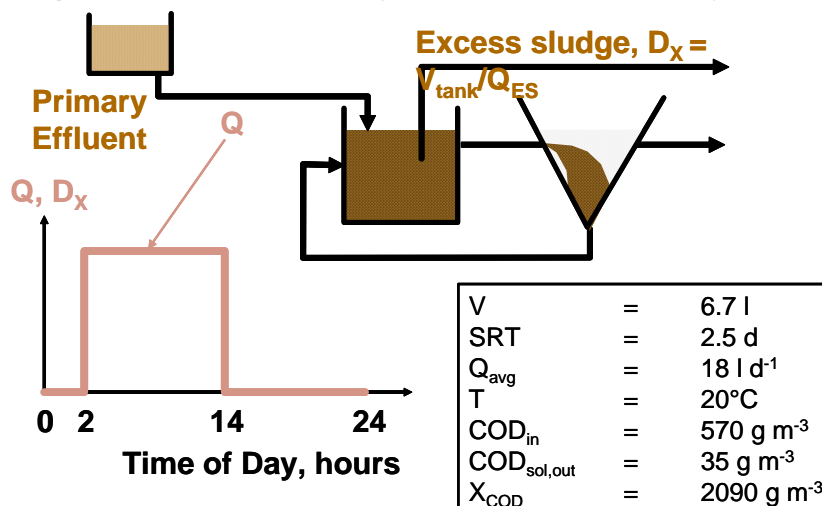
Growth rate $\rho = 1500 \text{ g}_{\text{BM}}/\text{d}$

Reaction rates: $r(\text{C}_6\text{H}_{12}\text{O}_6) = - 2.24 \cdot 1500 \text{ g/d}$
 $r(\text{CO}_2) = + 1.34 \cdot 1500 \text{ g/d}$

Components of a WWTP model

Development of a biokinetic model for AS plants

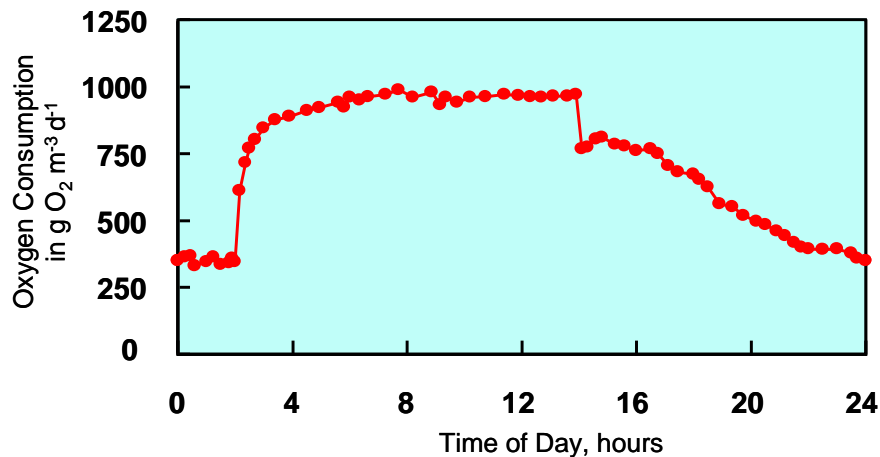
Pilot plant – characteristics (Ekama and Marais, 1978)



Components of a WWTP model

Development of a biokinetic model for AS plants

Pilot plant – Experimental results: (Ekama and Marais, 1978)



Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model 0 - Growth of heterotrophic organisms

| Component | SO ₂ | SI | SS | XH | Process rate r_i |
|----------------------|---------------------|----|------------------|----|--|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot X_H$ |

$Y_H = 0.67$ g COD(BM) / g COD(SS) ... Heterotrophic yield factor

$\mu_H = 4.0$ d⁻¹ ... max. heterotrophic growth rate

$K_{H,O_2} = 0.2$ g O₂/L ... Half-saturation coefficient for O₂

$K_{H,SS} = 5.0$ g COD(SS)/L ... Half-saturation coefficient for substrate

Influent: COD = 570 mg/L → SS = 540 mg/L; SI = 30 mg/L

Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model 0 - Growth of heterotrophic organisms

| Component | SO ₂ | SI | SS | XH | Process rate r_i |
|----------------------|---------------------|----|------------------|----|---|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |

$$\frac{d SO_2}{dt} = \left(1 - \frac{1}{Y_H}\right) \cdot \mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$$

$$\frac{d SS}{dt} = \left(-\frac{1}{Y_H}\right) \cdot \mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$$

$$\frac{d XH}{dt} = (1) \cdot \mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$$

Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model 0 - Growth of heterotrophic organisms

Components

1. Soluble oxygen SO₂
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. Heterotrophic biomass XH

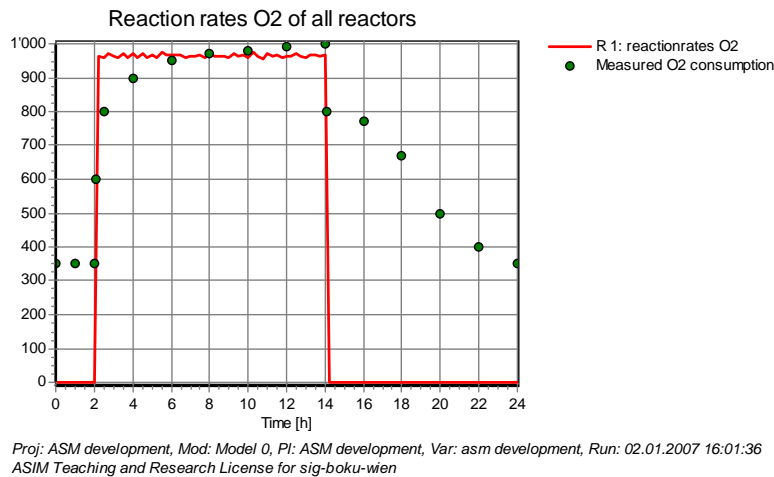
Processes

1. Aerobic growths of heterotrophic organisms

Components of a WWTP model

Development of a biokinetic model for AS plants

Model 0 - Comparison of simulated and measured oxygen consumption

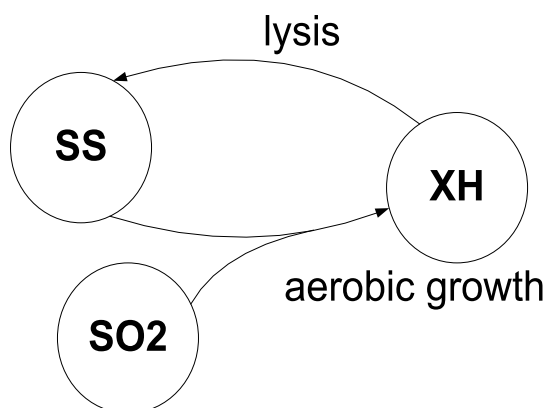


Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model A - Growth and Lysis of heterotrophic organisms



Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model A - Growth and Lysis of heterotrophic organisms

Components

1. Soluble oxygen SO₂
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. Heterotrophic biomass XH

Processes

1. Aerobic growths of heterotrophic organisms
2. **Lysis of heterotrophic organisms**

Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model A - Growth and Lysis of heterotrophic organisms

| Component | SO ₂ | SI | SS | XH | Process rate r_i |
|----------------------|---------------------|----|------------------|-----------|---|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |
| Lysis | | | +1 | -1 | $b_H \cdot XH$ |

$Y_H =$ 0.67 g COD(BM) / g COD(SS) ... Heterotrophic yield factor

$\mu_H =$ 2.5 d⁻¹ ... max. heterotrophic growth rate

$K_{H,O_2} =$ 0.2 g O₂/L ... Half-saturation coefficient for O₂

$K_{H,SS} =$ 5.0 g COD(SS)/L ... Half-saturation coefficient for substrate

$b_H =$ 0.4 d⁻¹ ... Heterotrophic decay constant

Influent: COD = 570 mg/L → SS = 540 mg/L; SI = 30 mg/L

Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model A - Growth and Lysis of heterotrophic organisms

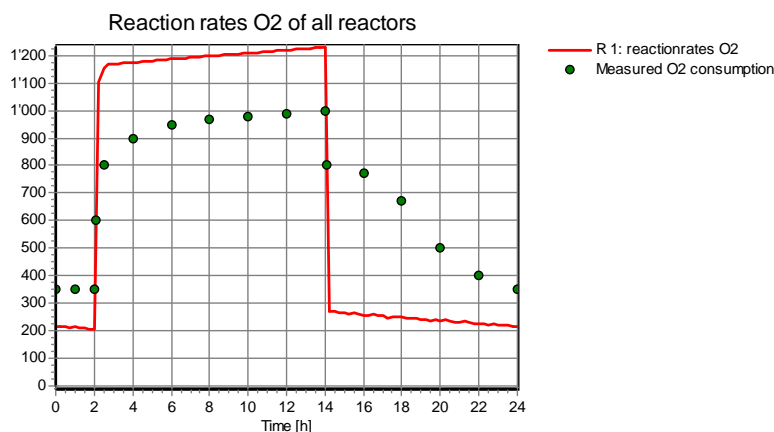
| Component | SO2 | SI | SS | XH | Process rate r_i |
|----------------------|---------------------|----|------------------|----|---|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |
| Lysis | | | +1 | -1 | $b_H \cdot XH$ |

$$\frac{dSS}{dt} = \left(-\frac{1}{Y_H}\right) \cdot \mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH \quad \underline{+1 \cdot b_H \cdot XH}$$

Components of a WWTP model

Development of a biokinetic model for AS plants

Model A - Comparison of simulated and measured oxygen consumption



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Components of a WWTP model

Development of a biokinetic model for AS plants

Aerobic degradation of soluble organic C

Model A - Growth and Lysis of heterotrophic organisms

Simulation results

- too rapid increase and decrease of oxygen consumption

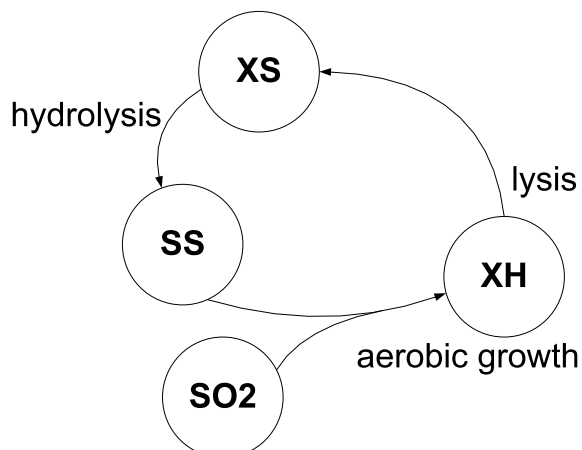
Extension

- higher molecular particulate organic substances XS that are transformed to low molecular solute organic substances SS due to hydrolysis

Components of a WWTP model

Development of a biokinetic model for AS plants

Model B - Aerobic degradation of soluble and particulate organic C



Components of a WWTP model

Development of a biokinetic model for AS plants

Model B - Aerobic degradation of soluble and particulate organic C

Components

1. Soluble oxygen SO₂
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. Heterotrophic biomass XH
5. **Slowly biodegradable particulate organic matter XS**

Processes

1. Aerobic growths of heterotrophic organisms
2. Lysis of heterotrophic organisms
3. **(Aerobic) Hydrolysis (transformation of XS to SS)**

Components of a WWTP model

Development of a biokinetic model for AS plants

Model B - Aerobic degradation of soluble and particulate organic C

| Component | SO ₂ | SI | SS | XH | XS | Process rate r_i |
|----------------------|---------------------|----|------------------|----|-----------|--|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |
| Lysis | | | | -1 | +1 | $b_H \cdot X_H$ |
| Hydrolysis | | | +1 | | -1 | $k_H \cdot \frac{XS/XH}{K_X + XS/XH} \cdot \frac{SO_2}{K_{Hyd,O_2} + SO_2} \cdot XH$ |

$k_H = 1.6 \text{ g COD(SS) / g COD(XS) ... Hydrolysis constant}$

$K_X = 0.04 \text{ g COD(SS) / g COD(BM) ... Saturation coefficient for adsorption}$

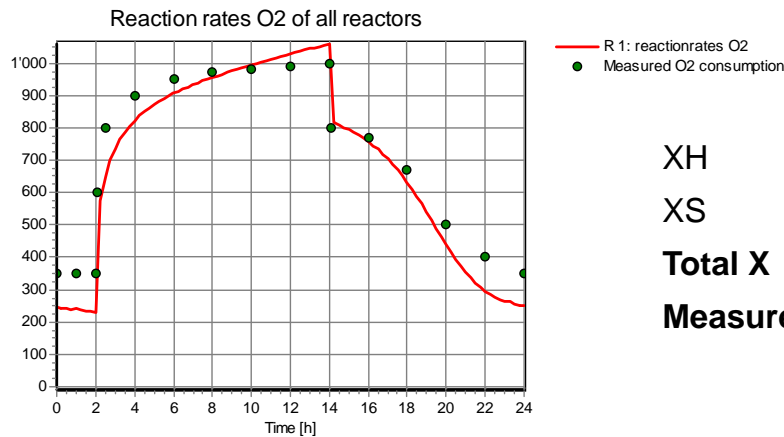
$K_{H,O_2} = 0.1 \text{ g O}_2 / \text{L ... Saturation coefficient for O}_2$

Influent: COD = 570 mg/L → XS = 430 mg/L; SS = 110 mg/L; SI = 30 mg/L

Components of a WWTP model

Development of a biokinetic model for AS plants

Model B - Comparison with measured oxygen consumption



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| | |
|------------------|------------------|
| XH | 1592 mg/L |
| XS | 25 mg/L |
| Total X | 1617 mg/L |
| Measured: | 2090 mg/L |

Components of a WWTP model

Development of a biokinetic model for AS plants

Model B - Aerobic degradation of soluble and particulate organic C

Simulation results

- oxygen consumption well modelled
- sludge production underestimated (Experiment 2.1 g COD / L; Simulation 1.6)

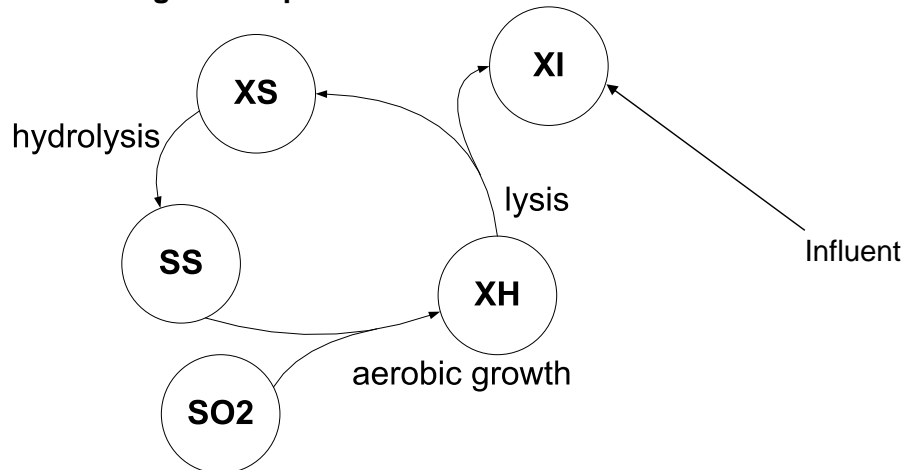
Extension

- inert particulate matter XI, that is not degradable and accumulates in the sludge

Components of a WWTP model

Development of a biokinetic model for AS plants

Model C - Non-degradable particulate matter



Components of a WWTP model

Development of a biokinetic model for AS plants

Model C - Non-degradable particulate matter

Components

1. Soluble oxygen SO₂
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. Heterotrophic biomass XH
5. Slowly biodegradable particulate organic matter XS
6. **Non-degradable particulate matter XI**

Processes

1. Aerobic growths of heterotrophic organisms
2. Lysis of heterotrophic organisms
3. (Aerobic) Hydrolysis (transformation of XS to SS)

Components of a WWTP model

Development of a biokinetic model for AS plants

Model C - Non-degradable particulate matter

| Component | SO2 | SI | SS | XH | XS | XI | Process rate r_i |
|----------------------|---------------------|----|------------------|----|---------|-------|--|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | +1 | | | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |
| Lysis | | | | -1 | $1-f_i$ | f_i | $b_H \cdot XH$ |
| Hydrolysis | | | +1 | | -1 | | $k_H \cdot \frac{XS/XH}{K_X + XS/XH} \cdot \frac{SO_2}{K_{Hyd,O_2} + SO_2} \cdot XH$ |

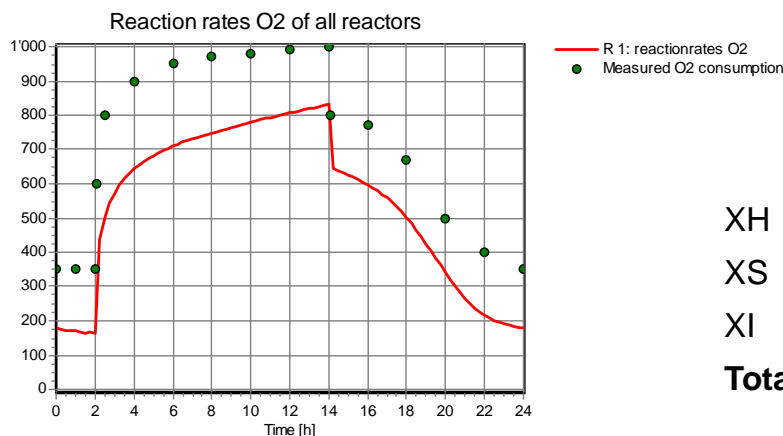
$f_i = 0.08$ g COD(XI)/g COD(BM) ... Inert matter generation during lysis

Influent: COD = 570 mg/L → XI = 100 mg/L; XS = 330 mg/L; SS = 110 mg/L; SI = 30 mg/L

Components of a WWTP model

Development of a biokinetic model for AS plants

Model C - Comparison of simulated and measured oxygen consumption



XH 1241 mg/L
XS 17 mg/L
XI 742 mg/L
Total X 2000 mg/L

Proj: ASM development, Mod: Model C, Pl: ASM development, Var: asm development, Run: 02.01.2007 16:25:45
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Components of a WWTP model

Development of a biokinetic model for AS plants

Model C - Non-degradable particulate matter

Simulation results

- sludge concentration fits
- oxygen consumption too low

Extension

- consideration of oxygen consumption due to nitrification
- effluent of experimental plant: $20 \text{ mg}_{\text{NO}_3\text{-N}}/\text{l}$ (mean value) → nitrification occurred although sludge age only 2.5 days

Components of a WWTP model

Development of a biokinetic model for AS plants

Model D - Nitrification

Components

1. Soluble oxygen SO_2
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. **Soluble Ammonia SNH_4**
5. **Soluble Nitrate (and Nitrite) SNO_3**
6. Heterotrophic biomass XH
7. Slowly biodegradable particulate organic matter XS
8. Non-degradable particulate matter XI
9. **Autotrophic biomass XA**

Components of a WWTP model

Development of a biokinetic model for AS plants

Model D - Nitrification

Processes

1. Aerobic growths of heterotrophic organisms
2. Lysis of heterotrophic organisms
3. (Aerobic) Hydrolysis (transformation of XS to SS)
4. **Aerobic growth of autotrophic organisms (Nitrifiers)**
5. **Lysis of autotrophic organisms**

Components of a WWTP model

Development of a biokinetic model for AS plants

Model D – Nitrification: Stoichiometry

| Process/Component | SO ₂ | SI | SS | SNH | SNO | XH | XS | XI | XA |
|---------------------------|--------------------------|----|------------------|---------------------------------|-----------------|----|------------------|----------------|----|
| Heterotrophic growth | $1 - \frac{1}{Y_H}$ | | $-\frac{1}{Y_H}$ | $i_N \cdot \frac{1 - Y_H}{Y_H}$ | | +1 | | | |
| Heterotrophic lysis | | | | | | -1 | 1-f _l | f _l | |
| Hydrolysis | | | +1 | | | | -1 | | |
| Autotrophic growth | $\frac{Y_A - 4.57}{Y_A}$ | | | $-i_N - \frac{1}{Y_A}$ | $\frac{1}{Y_A}$ | | | | +1 |
| Autotrophic lysis | | | | | | | 1-f _l | f _l | -1 |

Components of a WWTP model

Development of a biokinetic model for AS plants

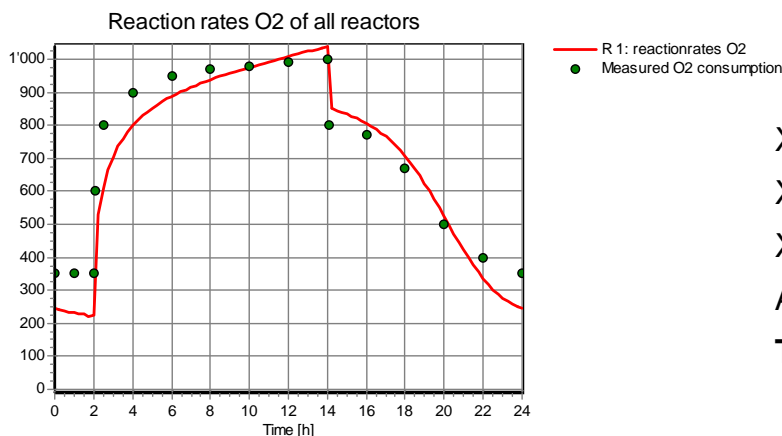
Model D – Nitrification: Kinetics

| Process | Process rate |
|---|--|
| Heterotrophic growth | $\mu_H \cdot \frac{SO_2}{K_{H,O_2} + SO_2} \cdot \frac{SS}{K_{H,SS} + SS} \cdot XH$ |
| Heterotrophic lysis | $b_H \cdot XH$ |
| Hydrolysis | $k_H \cdot \frac{XS/XH}{K_X + XS/XH} \cdot \frac{SO_2}{K_{Hyd,O_2} + SO_2} \cdot XH$ |
| Autotrophic growth (nitrification) | $\mu_A \cdot \frac{SO_2}{K_{A,O_2} + SO_2} \cdot \frac{SNH}{K_{A,SNH} + SNH} \cdot XA$ |
| Autotrophic lysis | $b_A \cdot XA$ |

Components of a WWTP model

Development of a biokinetic model for AS plants

Model D - Comparison of simulated and measured oxygen consumption



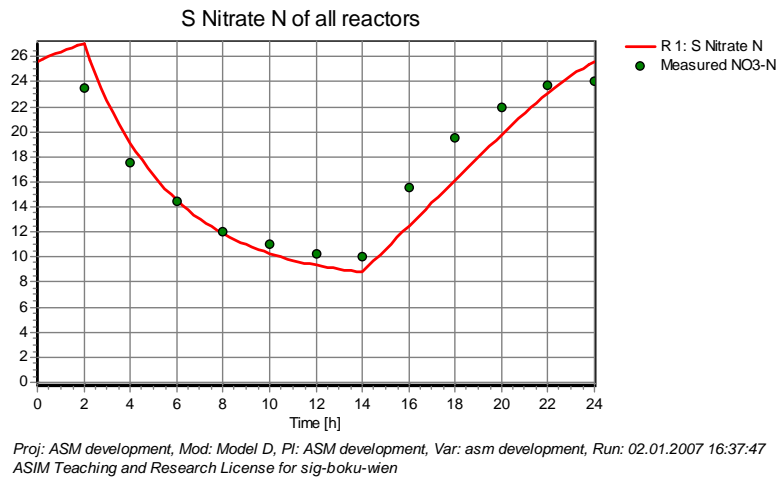
| | |
|----------------|------------------|
| XH | 1124 mg/L |
| XS | 17 mg/L |
| XI | 769 mg/L |
| AUT | 721 mg/L |
| Total X | 1931 mg/L |

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Components of a WWTP model

Development of a biokinetic model for AS plants

Model D - Comparison of simulated and measured effluent nitrate concentration



Components of a WWTP model

Development of a biokinetic model for AS plants

Model E – Denitrification

Components

1. Soluble oxygen SO₂
2. Soluble biodegradable organic matter SS
3. Soluble inert organic matter SI
4. Soluble Ammonia SNH₄
5. Soluble Nitrate (and Nitrite) SNO₃
6. Heterotrophic biomass XH
7. Slowly biodegradable particulate organic matter XS
8. Non-degradable particulate matter XI
9. Autotrophic biomass XA

Components of a WWTP model

Development of a biokinetic model for AS plants

Model E – Denitrification

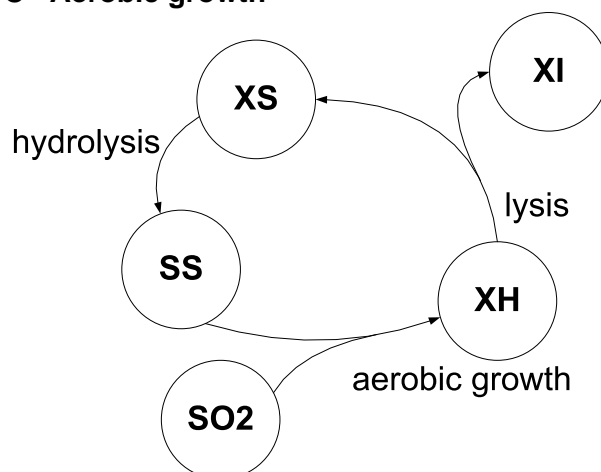
Processes

1. Aerobic growths of heterotrophic organisms
2. Lysis of heterotrophic organisms
3. (Aerobic) Hydrolysis (transformation of XS to SS)
4. Aerobic growth of autotrophic organisms (Nitrifiers)
5. Lysis of autotrophic organisms
6. **Anoxic growths of heterotrophic organisms (denitrification)**
7. **(Anoxic) Hydrolysis**

Components of a WWTP model

Development of a biokinetic model for AS plants

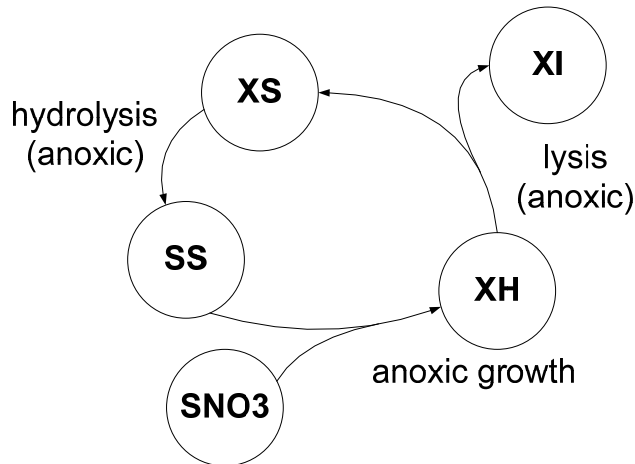
Model C - Aerobic growth



Components of a WWTP model

Development of a biokinetic model for AS plants

Model E – Denitrification



Components of a WWTP model

Development of a biokinetic model for AS plants

Model E – Denitrification

| Stoff | i | S_3 | S_4 | S_5 | X_6 | X_7 |
|---|-----|-----------------|-------------------------------|----------------------|-------|-------|
| Heterotrophic anoxic growth (denitrification) | | $\frac{1}{Y_H}$ | $i_N \cdot \frac{1-Y_H}{Y_H}$ | $\frac{-1+Y_H}{Y_H}$ | +1 | |
| Anoxic hydrolysis | | +1 | | | | -1 |

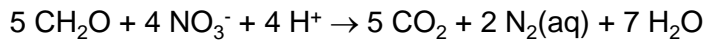
| j Prozess | Prozessgeschwindigkeit ρ_j |
|---|---|
| 6 Heterotrophic anoxic growth (denitrification) | $(\mu_{mH} \cdot \theta_w) \cdot \frac{K_{OH}}{K_{OH}+S_O} \cdot \frac{S_S}{K_S+S_S} \cdot \frac{S_{NO}}{K_{NO}+S_{NO}} \cdot X_H$ |
| 7 Anoxic hydrolysis | $(k_h \cdot \theta_h) \cdot \frac{X_S/X_H}{K_X+X_S/X_H} \cdot \frac{K_{OH}}{K_{OH}+S_O} \cdot \frac{S_{NO}}{K_{NO}+S_{NO}} \cdot X_H$ |

Components of a WWTP model

Development of a biokinetic model for AS plants

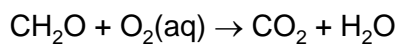
Model E – Denitrification – Stoichiometry

Denitrification (C-source = CH₂O = formaldehyde):



$$(4 \times 14) = 56\text{g NO}_3\text{-N for } (5 \times 30) = 150\text{g CH}_2\text{O} \rightarrow 2.679 \text{ g}_{\text{CH}_2\text{O}} / \text{g}_{\text{NO}_3\text{N}}$$

Oxidation:



$$32\text{g O}_2 \text{ for } 30\text{g CH}_2\text{O} \rightarrow 1.067\text{g}_{\text{O}_2} / \text{g}_{\text{CH}_2\text{O}}$$

$$2.679 \times 1.067 = \mathbf{2.86 \text{ g}_{\text{O}_2} / \text{g}_{\text{NO}_3\text{N}}}$$

Components of a WWTP model

Development of a biokinetic model for AS plants

Models for phosphorus removal

1) Biological P-Removal (Bio-P)

- „Luxury Uptake“ (e.g. Acinetobacter)
- XPAO = P-accumulating organisms

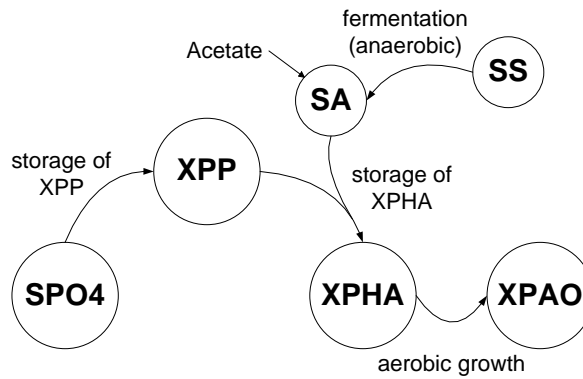
2) Chemical P-Precipitation

- $\text{Me}(\text{PO}_4)_x(\text{OH})_{3-3x} \dots \text{Me} \dots \text{metal}$
- $1/x = b \dots$ Stoichiometric ratio for dosing of Me

Components of a WWTP model

Development of a biokinetic model for AS plants

Model for Bio-P (EBPR – Enhanced Biological Phosphorus Removal) e.g. ASM2



XPHA ... cell-internal organic storage product of XPAO (Poly-hydroxy-alkanoates)

XPP ... cell-internal inorganic storage product of XPAO (Poly-Phosphat)

Components of a WWTP model

Development of a biokinetic model for AS plants

Model for chemical P precipitation (chemical equilibrium)

Components

- Soluble phosphorus (orthophosphate) SPO4
- Soluble Iron SFe
- Iron-hydroxide XFeOH
- Iron-phosphate XFeP

Processes

- Oxidation $\text{Fe}^{2+} \rightarrow \text{Fe}^{3+}$
- Precipitation $\text{SPO}_4 \rightarrow \text{XFeP}$
- Re-dissolution $\text{XFeP} \rightarrow \text{SPO}_4$

Components of a WWTP model

Development of a biokinetic model for AS plants

Model F - Model for chemical P precipitation

| Component | soluble P | Fe ²⁺ | Fe(OH) ₃ | FePO ₄ | O ₂ |
|----------------|----------------|------------------|---------------------|-------------------|----------------|
| i | S _P | S _{Fe} | X _{FH} | X _{FP} | S _O |
| | 10 | 11 | 12 | 13 | 1 |
| Oxidation* | | -1 | 1.91 | | -0.14 |
| Precipitation | -1 | | -3.45 | 4.87 | |
| Re-dissolution | 1 | | 3.45 | -4.87 | |

| j | Processes | Process rate | ρ _j |
|----|----------------|----------------------------------|----------------|
| 8 | Oxidation* | $K_O \cdot S_{Fe} \cdot S_{O_2}$ | |
| 9 | Precipitation | $K_F \cdot S_P \cdot X_{FH}$ | |
| 10 | Re-dissolution | $K_R \cdot X_{FP}$ | |

Components of a WWTP model

IWA Activated Sludge Models

IWA – International Water Association (since 2000)

IAWQ – International Association on Water Quality (1990-2000)

IAWPRC – International Association on Water Pollution Research and Control (<1990)

Components of a WWTP model

IWA Activated Sludge Models

IWA Activated Sludge Model No.1 (ASM1, 1987)

Fundamentals

- 13 components and 8 processes
- Describes oxygen consumption, sludge production and nitrogen elimination
- Mass balances for C and N
- Standard model for numerical simulation of activated sludge plants

References

- Henze, M., et al. (1987): Activated Sludge Model No.1. In: IWA Scientific and Technical Report No.9 (2000), pp. 5-37.
- Bornemann, C., et al. (1998): Hinweise zur dynamischen Simulation von Belebungsanlagen mit dem Belebtschlammmodell Nr.1 der IAWQ. *Korrespondenz Abwasser* 45(3), 455-482.

Components of a WWTP model

IWA Activated Sludge Models

ASM1 – components

1. SI: Soluble inert organic matter
2. SS: Soluble biodegradable organic matter
3. XS: Slowly biodegradable particulate organic matter
4. XI: Non-degradable particulate matter
5. XBH: Heterotrophic biomass
6. XBA: Autotrophic biomass
7. **XP: Particulate products of decay processes**
8. SO: Soluble oxygen
9. SNO: Soluble nitrate and nitrite nitrogen
10. SNH: Soluble ammonia nitrogen
11. **SND: Soluble organic nitrogen**
12. **XND: Particulate organic nitrogen**
13. **SALK: Alkalinity**

Components of a WWTP model

IWA Activated Sludge Models

ASM1 – processes

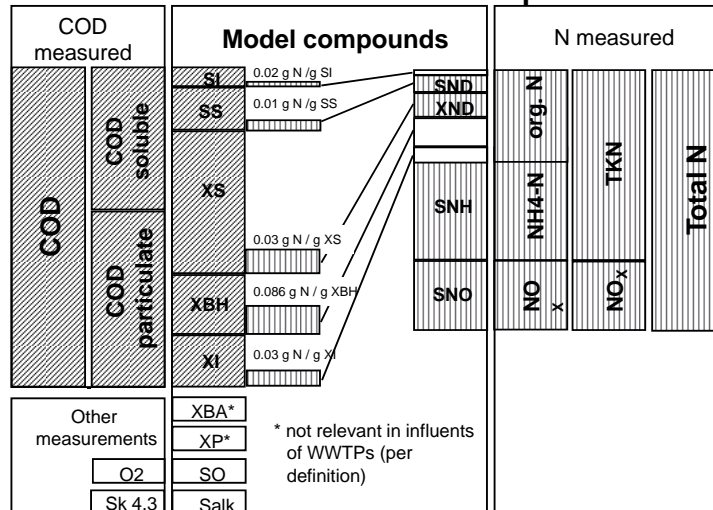
1. Aerobic growths of heterotrophic organisms
2. Anoxic growths of heterotrophic organisms (denitrification)
3. Aerobic growth of autotrophic organisms (nitrification)
4. Lysis of heterotrophic organisms
5. Lysis of autotrophic Organisms
6. **Ammonification**
7. Aerobic Hydrolysis
8. Anoxic Hydrolysis

| j | Component → Process ↓ | i | | | | | | | | | | | | | Process Rate, ρ_j [$ML^{-3}T^{-1}$] | | |
|--|--|---|--|---|---|---|---|--|---|---|---|---|---|--------------------------|---|--|---|
| | | 1 S_1 | 2 S_2 | 3 X_1 | 4 X_2 | 5 $X_{B,H}$ | 6 $X_{B,A}$ | 7 X_P | 8 S_{O_2} | 9 S_{NO_3} | 10 S_{NH_4} | 11 S_{ND} | 12 X_{ND} | 13 S_{ALK} | | | |
| 1 | Aerobic growth of heterotrophs | | $\frac{1}{Y_H}$ | | | 1 | | | $-\frac{1-Y_H}{Y_H}$ | | | | | | | $-\frac{i_{XB}}{14}$ | $\mu_H \left(\frac{S_2}{K_S + S_2} \right) \left(\frac{S_{O_2}}{K_{O,H} + S_{O_2}} \right) X_{B,H}$ |
| 2 | Anoxic growth of heterotrophs | | $\frac{1}{Y_H}$ | | | 1 | | | $-\frac{1-Y_H}{2.86 Y_H}$ | | | | | | | $-\frac{i_{XB}}{14}$ | $\mu_H \left(\frac{S_2}{K_S + S_2} \right) \left(\frac{K_{O,H}}{K_{O,H} + S_{O_2}} \right) \times \left(\frac{S_{NO_3}}{K_{NO_3} + S_{NO_3}} \right) \eta_R X_{B,H}$ |
| 3 | Aerobic growth of autotrophs | | | | | | 1 | | $-\frac{4.57 - Y_A}{Y_A}$ | $\frac{1}{Y_A}$ | | | | | | $-\frac{i_{XB}}{14} - \frac{1}{7 Y_A}$ | $\mu_A \left(\frac{S_{NH_4}}{K_{NH_4} + S_{NH_4}} \right) \left(\frac{S_{O_2}}{K_{O,A} + S_{O_2}} \right) X_{B,A}$ |
| 4 | 'Decay' of heterotrophs | | | | $1 - f_P$ | -1 | | f_P | | | | | | | | | $b_H X_{B,H}$ |
| 5 | 'Decay' of autotrophs | | | | $1 - f_P$ | -1 | f_P | | | | | | | | | | $b_A X_{B,A}$ |
| 6 | Ammonification of soluble organic nitrogen | | | | | | | | | | | 1 | -1 | | $\frac{1}{14}$ | | $k_A S_{ND} X_{B,H}$ |
| 7 | 'Hydrolysis' of entrapped organics | | 1 | | -1 | | | | | | | | | | | | $k_h \frac{X_B/X_{B,H}}{K_X + (X_B/X_{B,H})} \left[\left(\frac{S_{O_2}}{K_{O,H} + S_{O_2}} \right) + \eta_h \left(\frac{K_{O,H}}{K_{O,H} + S_{O_2}} \right) \left(\frac{S_{NO_3}}{K_{NO_3} + S_{NO_3}} \right) \right] X_{B,H}$ |
| 8 | 'Hydrolysis' of entrapped organic nitrogen | | | | | | | | | | | | 1 | -1 | | | $\rho_8 (X_{ND}/X_B)$ |
| Observed Conversion Rates [$ML^{-3}T^{-1}$] | | $r_i = \sum_j \nu_{ij} \rho_j$ | | | | | | | | | | | | | | | |
| Stoichiometric Parameters: Heterotrophic yield: Y_H Autotrophic yield: Y_A Fraction of biomass yielding particulate products: f_P Mass N/Mass COD in biomass: i_{XB} Mass N/Mass COD in products from biomass: i_{XP} | | Soluble inert organic matter [$M(COD)L^{-3}$] | Readily biodegradable substrate [$M(COD)L^{-3}$] | Particulate inert organic matter [$M(COD)L^{-3}$] | Slowly biodegradable substrate [$M(COD)L^{-3}$] | Active heterotrophic biomass [$M(COD)L^{-3}$] | Active autotrophic biomass [$M(COD)L^{-3}$] | Particulate products arising from biomass decay [$M(COD)L^{-3}$] | Oxygen (negative COD) [$M(-COD)L^{-3}$] | Nitrate and nitrite nitrogen [$M(N)L^{-3}$] | NH_4^+ + NH_3 nitrogen [$M(N)L^{-3}$] | Soluble biodegradable organic nitrogen [$M(N)L^{-3}$] | Particulate biodegradable organic nitrogen [$M(N)L^{-3}$] | Alkalinity – Molar units | Kinetic Parameters: Heterotrophic growth and decay: $\mu_H, K_S, K_{O,H}, K_{NO_3}, b_H$ Autotrophic growth and decay: $\mu_A, K_{NH_4}, K_{O,A}, b_A$ Correction factor for anoxic growth of heterotrophs: η_R Ammonification: k_A Hydrolysis: k_h, K_X Correction factor for anoxic hydrolysis: η_h | | |

Components of a WWTP model

IWA Activated Sludge Models

ASM1 – measured concentrations vs. model compounds



Components of a WWTP model

IWA Activated Sludge Models

ASM1 – limitations

1. a constant value of pH,
2. constant coefficients in the rate equations,
3. constant stoichiometric factors, and
4. a limited temperature range from 10°C to 25°C.

Components of a WWTP model

IWA Activated Sludge Models

IWA Activated Sludge Model No.2 (ASM2, 1995)

- ASM1 (C+N) + Bio-P + chemical P-precipitation
- org. N + org. P modelled as part of COD
- 19 components und 19 processes

IWA Activated Sludge Model No.2d (ASM2d, 1999)

- ASM2 + more accurate Bio-P Model
- 19 components und 21 processes

Components of a WWTP model

IWA Activated Sludge Models

IWA Activated Sludge Model No.3 (ASM3, 1999)

Fundamentals

- new C-degradation in relation to ASM1 (cell-internal storage products)
- without P (extensions for P are existing)
- 13 components and 12 processes

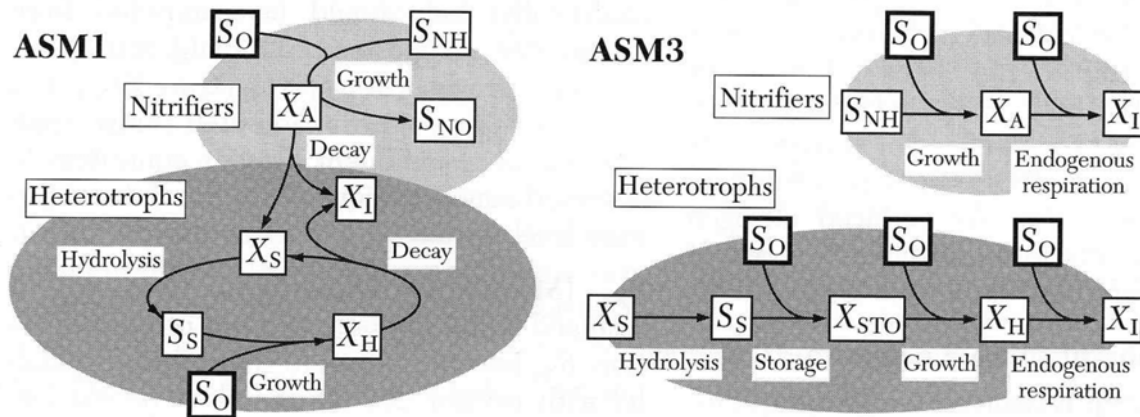
References

- Gujer, W., et al. (1999): Activated Sludge Model No.3. *Water Science & Technology* 39(1), 183-193 .
- Koch, G., et al. (2000): Calibration and validation of Activated Sludge Model No.3 for Swiss municipal wastewater. *Water Research* 34(14), 3580-3590.
- Rieger, L., et al. (2001): The EAWAG Bio-P module for Activated Sludge Model No.3. *Water Research* 35(16), 3887-3903.

Components of a WWTP model

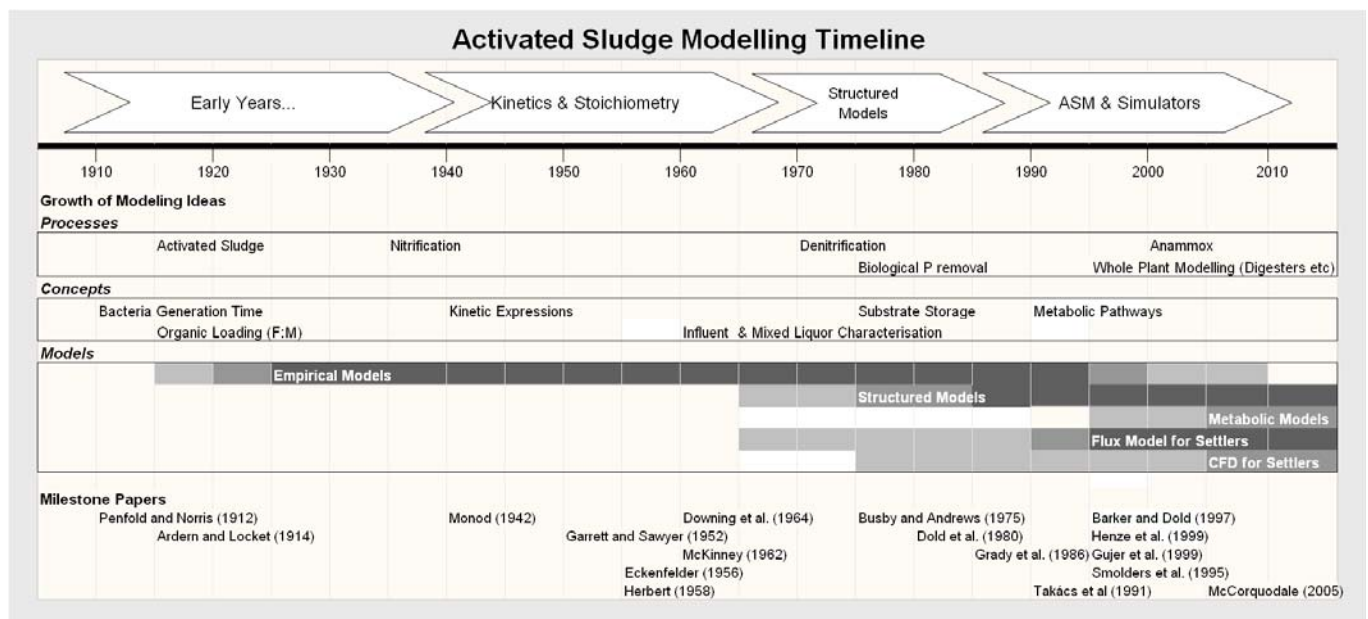
IWA Activated Sludge Models

Differences ASM1 - ASM3



Components of a WWTP model

Development of a biokinetic model for AS plants



Components of a WWTP model

References for AS modelling timeline

- Penfold, W. J., and Norris, D. (1912): The Relation of Concentration of Food Supply to the Generation Time of Bacteria. J. Hyg., 12, 527.
- Ardern, E. and Lockett W.T. (1914): Experiments on the oxidation of sewage without the aid of filters. J. Soc. Chem. Ind. 33, 523-39.
- Monod, J. (1942): Recherches sur la croissance des cultures bactériennes" Librairie scientifique, Paris, 1942 (In French)
- Garrett, M.T. Jr. and Sawyer, C.N. (1952): "Kinetics of Removal of Soluble Substrates by Activated Sludge." Proc. 7th Ind. Was. Conf., Purdue University, Lafayette, Indiana, p 51.
- Herbert, D. (1958): Recent progress in microbiology. VII International Congress for Microbiology. Ed. G Tunevall, Almquist and Wiksel, Stockholm, 381.
- Eckenfelder, W. W. (1956): Studies on the Oxidation Kinetics of Biological Sludges. Sewage and Industrial Wastes 28 (8), 983-990.
- McKinney, R. E. (1962): Mathematics of Complete Mixing Activated Sludge. J. Sanit. Eng. Divl, Proc. Am. Soc Civ Eng 88, SA3.
- Downing AL, Painter HA and Knowles G (1964): Nitrification in the activated sludge process. J. Proc. Inst. Sewage Purif. 64, 130-153.
- Busby, J.B. and Andrews, J.F. (1975): Dynamic modeling and control strategies for the activated sludge process. JWPCFA, 47, 5, p 1055-1080.
- Dold, P.L., Ekama, G.A. and Marais, G.v.R. (1980): "A General Model for the Activated Sludge Process." Prog. Wat. Tech., 12, (Toronto) 44-77.
- Grady, C.P.L. Jr., Gujer, Willy, Henze, M., Marais, G.v.R., and Matsuo, T. (1986): "A Model for Single Sludge Wastewater Treatment Systems." Water Sci Technol 18, 47-61
- Takács I., G.G. Patry and D. Nolasco (1991). A dynamic model of the clarificationthickening process. Water Res. 25(10), 1263-1271.
- Smolders, G.J.F., Van-Der-Meij, J., van Loosdrecht, M.C.M. and Heijnen, J.J. (1995): A structured metabolic model for anaerobic and aerobic stoichiometry and kinetics of the biological phosphorus removal process. Biotechnology and Bioengineering 47(3), 277-287
- Barker, P.S. and P. L. Dold (1997): General model for biological nutrient removal activated sludge systems: model presentation. Water Environ Res 69(5), 969-984.
- Henze, M., Gujer, W., Mino, T., Matsuo, T., Wentzel, M.C., Marais, G.V.R. and van Loosdrecht, M.C.M. (1999): Activated Sludge Model No.2d, ASM2d. Water Sci Technol 39(1), 165-182.
- Gujer, W., Henze, M., Mino, T. and van Loosdrecht, M.C.M. (1999). Activated Sludge Model No. 3. Water Sci Technol 39(1), 183-193.
- McCorquodale, A., Griborio, A., Georgiou, I. (2005). A Public Domain Settling Tank Model. Proceedings WEFTEC 2005, WEF, Alexandria, VA, USA.

Components of a WWTP model

Development of a biokinetic model for AS plants

Brief history of AS modelling

Before 1983: different approaches: C and N removal

- Empirical design
- Early models based on BOD, MLSS
- Experiments at University of Capetown (UCT) from 60s

1983 : IWA TG "Mathematical modelling for design and operation of biological wastewater treatment"

- ASM1 Henze *et al.*, 1986 (Activated sludge Model n°1)

1995/1999: Bio-P removal included in IWA model series ASM2/2d

1997 : C/N/P model (Barker and Dold,1997)

Since then:

- New concepts C and N removal (ASM3), ASM3+Bio-P (EAWAG), metabolic model (Delft Univ.), UCTPHO+, elemental balances (Takács *et al.*, 2006), ...

Components of a WWTP model

Development of a biokinetic model for AS plants

Available biokinetic models

C and N removal

- ASM1
- ASM3, ASM3C

C/N and P removal

- ASM2, ASM2d, ASM3+Bio-P
- Barker & Dold
- UCTPHO+
- ASDM
- TUDelft

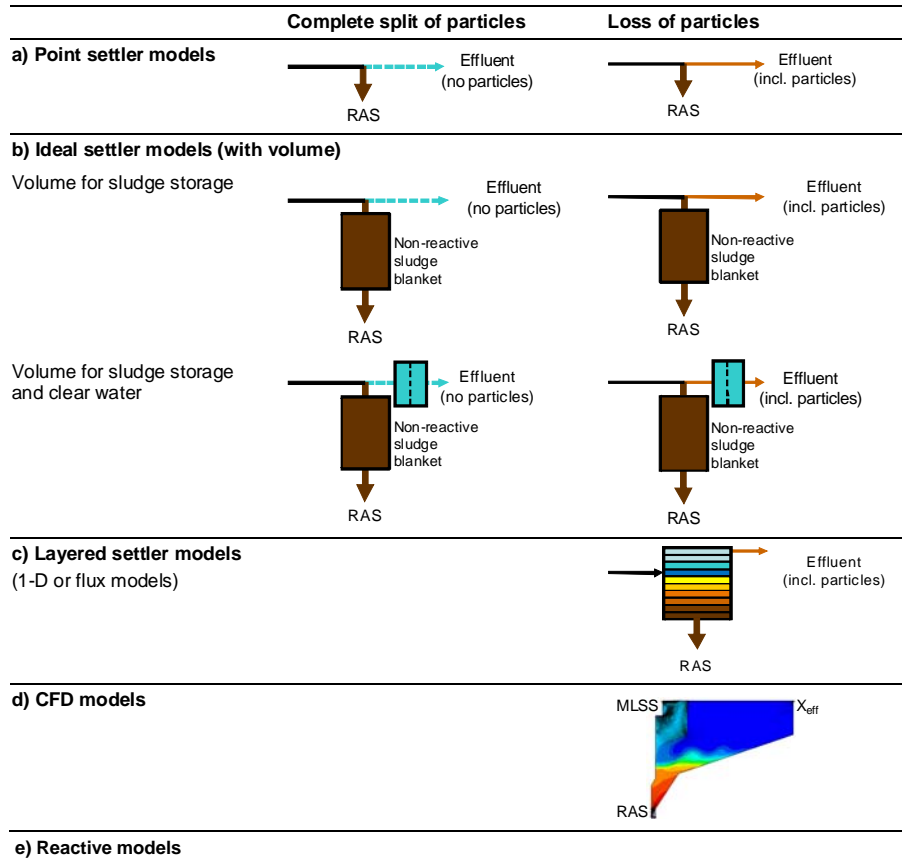
many others, modified models...

Components of a WWTP model

Models for secondary clarifiers

Secondary clarifiers

- Separation activated sludge / effluent
- Settling and thickening of AS
- Sludge storage in the case of rainfall events in mixed systems

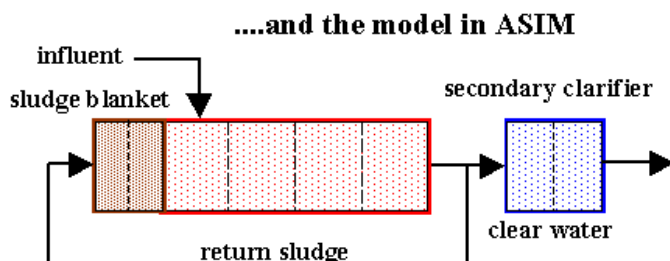
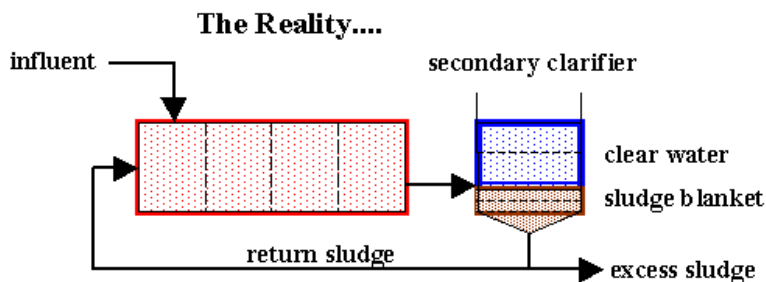


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Components of a WWTP model

Models for secondary clarifiers



Secondary clarifier model in ASIM



University of Natural Resources and Life Sciences, Vienna
Department of Water, Atmosphere, and Environment

Components of a WWTP model

Models for secondary clarifiers

1D layered settler models

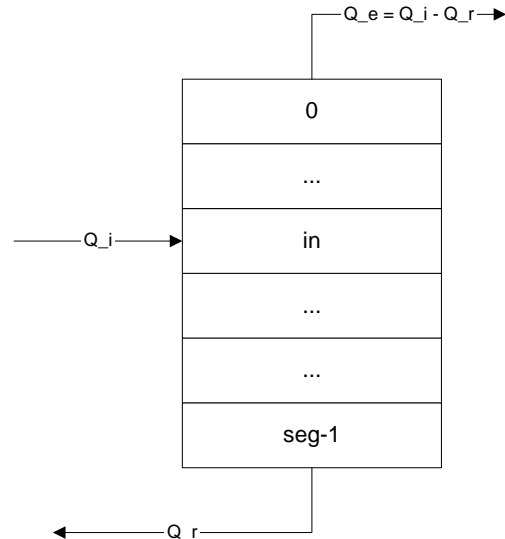
vertical transport due to

1. convection (with flowing water) and
2. sedimentation (relative to flowing water)

Q_i = influent

Q_r = recirculation

Q_e = effluent = $Q_i - Q_r$



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Models for secondary clarifiers

1D layered settler models

Settling velocity

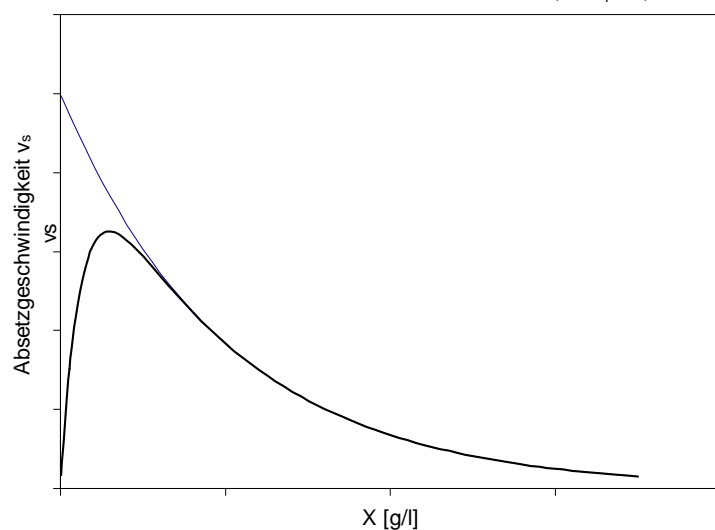
Most simple:

$$v_s = v_{s0} \cdot e^{-nX}$$

$X < \rightarrow v_s$ too high

Double exponential form

$$v_s = v_{s0} \cdot e^{-nX} - v_{s0} \cdot e^{-mX}$$





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Modelling exercise 1

Dynamic simulation of a AS plant

Introduction to ASIM

Objectives

1. Implementation of a AS plant into the ASIM simulation software
2. Optimisation on nitrogen elimination



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